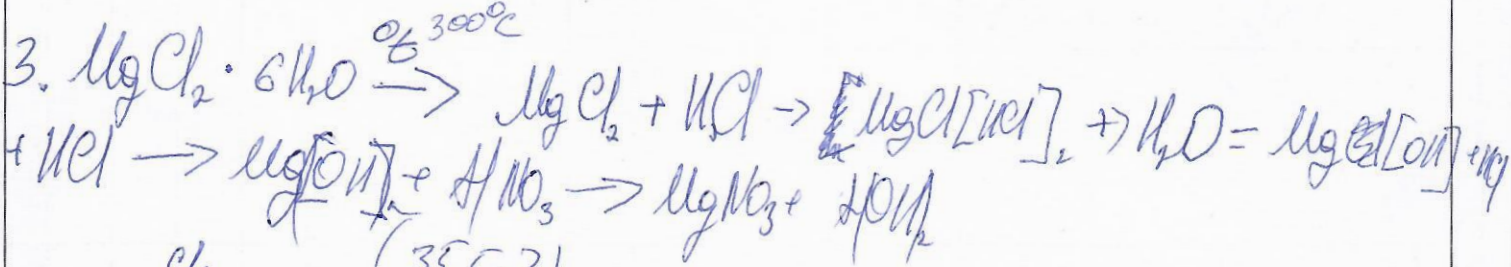
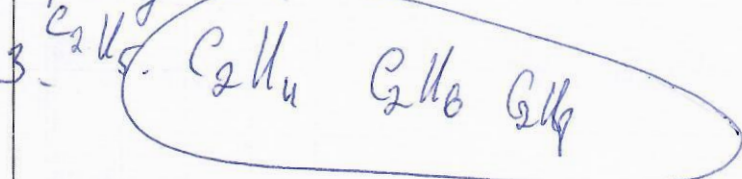
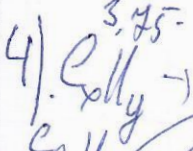


покажется в 4% раствор $K[OH]_2$

2. $\frac{A}{B}$
B



$M(Mg) = 39 + (16 \cdot 2) = 71$
 $M(MgCl_2) = 39 + (35.5 \cdot 2) = 105$
 $w = \frac{105}{0.291} = 360.83$



4-3-с